

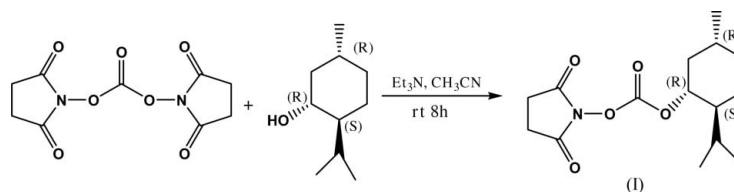
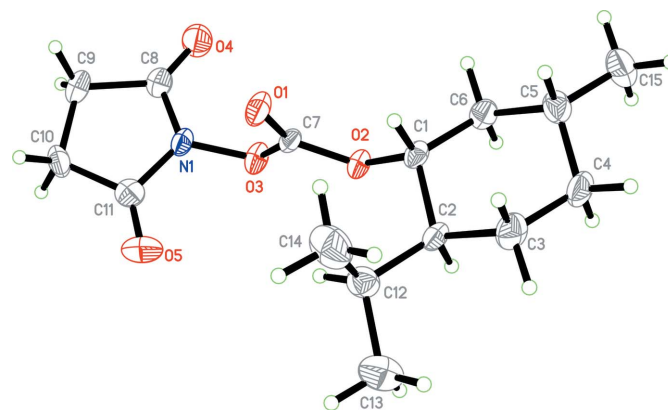
Tong-Jian Wang,<sup>a</sup> Hua Fang,<sup>a</sup>  
Fang Cheng,<sup>b</sup> Guo Tang<sup>a\*</sup> and  
Yu-Fen Zhao<sup>c</sup><sup>a</sup>Department of Chemistry, The Key Laboratory for Chemical Biology of Fujian Province, College of Chemistry and Chemical Engineering, Xiamen University, Xiamen 361005, People's Republic of China, <sup>b</sup>Key Laboratory of Analytical Sciences, Ministry of Education, and Department of Chemistry, College of Chemistry and Chemical Engineering, Xiamen University, Xiamen 361005, People's Republic of China, and <sup>c</sup>Department of Pharmaceutical Science, Medical College, Xiamen University, Xiamen 361005, People's Republic of China

Correspondence e-mail: t12g21@xmu.edu.cn

**Key indicators**Single-crystal X-ray study  
 $T = 293\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$   
 $R$  factor = 0.062  
 $wR$  factor = 0.126  
Data-to-parameter ratio = 8.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.**2,5-Dioxopyrrolidin-1-yl (1*R*,2*S*,5*R*)-2-isopropyl-5-methylcyclohexyl carbonate**The title compound,  $\text{C}_{15}\text{H}_{23}\text{NO}_5$ , was obtained as colourless crystals by the reaction of *N,N'*-disuccinimidyl carbonate (DSC) and *L*-(-)-menthol. The molecular packing is stabilized by weak  $\text{C}-\text{H}\cdots\text{O}$  hydrogen-bonding interactions.

Received 14 November 2006

Accepted 20 November 2006

**Comment**The title compound, (I), is a key intermediate used in the synthesis of a series of biologically active polyfunctional molecules for enzyme active sites (Ghosh *et al.*, 1992). It can react with various chiral alcohols and amines to yield active esters of interest in medicinal chemistry. The absolute configuration of the three stereocentres remains unchanged during the synthetic procedure. The molecular packing is stabilized by weak  $\text{C}-\text{H}\cdots\text{O}$  hydrogen-bonding interactions (Table 1).**Experimental***N,N'*-Disuccinimidyl carbonate (DSC), was prepared according to the procedure of Pereira *et al.* (1998). To a solution of DSC (1.5 mmol, 0.384 g) and *L*-(-)-menthol (1.0 mmol, 0.156 g) in dried acetonitrile (10 ml) at room temperature was added triethylamine (3 mmol) dropwise. The resulting mixture was stirred at room temperature**Figure 1**

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are represented by spheres of arbitrary radii.

until no starting L-(–)-menthol remained, as monitored by thin-layer chromatography (8 h). The mixture was concentrated under reduced pressure and the residue was diluted with a saturated aqueous NaHCO<sub>3</sub> solution (10 ml) and extracted thoroughly with ethyl acetate (3 × 20 ml). The combined extracts were washed with a saturated aqueous NaCl solution (8 ml) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation of the solvent provided the mixed carbonate (I) which was purified by column chromatography over silica gel (petroleum ether/ethyl acetate 5:1 v/v) to obtain a white solid. Single crystals were obtained by recrystallization from a mixture of petroleum ether/ethyl acetate (5:1 v/v).

#### Crystal data

C <sub>15</sub> H <sub>23</sub> NO <sub>5</sub>	Z = 4
M <sub>r</sub> = 297.34	D <sub>x</sub> = 1.237 Mg m <sup>-3</sup>
Orthorhombic, P <sub>2</sub> <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	Mo Kα radiation
a = 6.5911 (14) Å	μ = 0.09 mm <sup>-1</sup>
b = 7.7815 (17) Å	T = 293 (2) K
c = 31.125 (7) Å	Block, colourless
V = 1596.4 (6) Å <sup>3</sup>	0.40 × 0.27 × 0.20 mm

#### Data collection

Bruker APEX area-detector diffractometer	7845 measured reflections
φ and ω scans	1652 independent reflections
Absorption correction: multi-scan (SADABS; Bruker, 2001)	1621 reflections with I > 2σ(I)
T <sub>min</sub> = 0.964, T <sub>max</sub> = 0.982	R <sub>int</sub> = 0.028
	θ <sub>max</sub> = 25.0°

#### Refinement

Refinement on F <sup>2</sup>	w = 1/[σ <sup>2</sup> (F <sub>o</sub> <sup>2</sup> ) + (0.0276P) <sup>2</sup> + 0.8697P]
R[F <sup>2</sup> > 2σ(F <sup>2</sup> )] = 0.062	where P = (F <sub>o</sub> <sup>2</sup> + 2F <sub>c</sub> <sup>2</sup> )/3
wR(F <sup>2</sup> ) = 0.127	(Δ/σ) <sub>max</sub> = 0.004
S = 1.39	Δρ <sub>max</sub> = 0.18 e Å <sup>-3</sup>
1652 reflections	Δρ <sub>min</sub> = -0.18 e Å <sup>-3</sup>
193 parameters	
H-atom parameters constrained	

**Table 1**

Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C1—H1···O5 <sup>i</sup>	0.98	2.51	3.460 (5)	164
C10—H10A···O1 <sup>ii</sup>	0.97	2.41	3.255 (5)	145

Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $x + \frac{1}{2}, -y + \frac{5}{2}, -z + 2$ .

All H atoms were placed in geometrically idealized positions and treated as riding on their parent atoms, with C—H = 0.98 (CH), 0.97 (CH<sub>2</sub>) or 0.96 Å (CH<sub>3</sub>), and U<sub>iso</sub>(H) = 1.2 U<sub>eq</sub>(CH and CH<sub>2</sub>) or 1.5 U<sub>eq</sub>(methyl C). In the absence of significant anomalous scattering effects, Friedel pairs were merged; the absolute configuration was assumed from the synthesis.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97.

The project was supported by the National Natural Science Foundation of China (No. 20572061) and we also thank the Key Laboratory for the Physical Chemistry of the Solid Surface for providing the X-ray diffraction facilities.

#### References

- Bruker (2001). SAINT (Version 6.22), SMART (Version 5.625) and SADABS (Version 2.03). Bruker AXS Inc., Madison, Wisconsin, USA.
- Pereria, D., Hai, T. T. & Nelson, D. D. (1998). *Synth. Commun.* **28**, 4019–4024.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Ghosh, A. K., Duong, T. T., Mckee, S. P. & Thompson, W. J. (1992). *Tetrahedron Lett.* **33**, 2781–2784.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.